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Note

Determination of L-glutamine in preparations using high-performance liquid chromatography

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L-Glutamine is an important medicine which has been used for the treatment of peptic ulcers, especially in Japan¹. The determination of L-glutamine has been carried out by means of, *e.g.*, ion-exchange chromatography²⁻⁹, gas chromatography¹⁰⁻¹³, fluorimetric methods^{14,15}, methods for measuring the amount of ammonia liberated by the actions of enzymes¹⁶⁻²¹ or acids^{22,23} on L-glutamine, and various other methods²⁴⁻²⁷. In this paper, the determination of L-glutamine in preparations using high-performance liquid chromatography is described.

EXPERIMENTAL

Reagents

0.02 *M* Potassium dihydrogen phosphate was prepared by weighing 2.72 g of potassium dihydrogen phosphate in sufficient water to make 1000 ml. 0.2 *M* Sodium hydroxide was prepared by weighing 0.8 g of sodium hydroxide in sufficient water to make 100 ml. Phosphate buffer solution was adjusted to pH 5.0 by mixing 0.02 *M* potassium dihydrogen phosphate and 0.2 *M* sodium hydroxide, filtered through a membrane filter (Toyo, TM-2, 0.45 μm ; Toyo Roshi, Tokyo, Japan) and used as a mobile phase after degassing. L-Glutamine (Ajinomoto, Tokyo, Japan) was dried at 80° for 3 h, *ca.* 50 mg accurately weighed and dissolved in sufficient water to make exactly 50 ml, and used as a standard solution. An amount of sample, equivalent to 30 mg of L-glutamine, was weighed accurately in a centrifuge tube with a stopper, 30 ml of water was added, it was shaken for 10 min, filtered through Toyo No. 5 filter paper 10 ml of first filtrate was removed, 5 ml of following clear filtrate was collected and used as a sample solution. All other reagents were commercially available products of analytical reagent grade.

Apparatus and procedure

The chromatograph consisted of an Altex pump (Model 110-A; Berkeley, Calif., U.S.A.) operated at a flow-rate of 0.8 ml/min. The column effluent was monitored by a Shimadzu UV variable-wavelength detector (Model 202; Kyoto, Japan) with a 20- μl flow-through cell. The detector was operated at 210 nm ($32 \cdot 10^{-2}$ a.u.f.s.). The column consisted of a stainless-steel tube (25 cm \times 4.6 mm I.D.) packed

with Nucleosil 10 SB (Macherey, Nagel & Co., Düren, G.F.R.) by the balanced viscosity and dispersion slurry-packing technique. All experiments were carried out at room temperature. A 20- μ l volume of standard or sample solution was injected accurately by using a syringe loading sample injector (Model 7120; Rheodyne, Berkeley, Calif., U.S.A.). The chromatogram was recorded at a chart speed of 5 mm/min. The L-glutamine content in the sample was calculated by peak height measurements relative to standard solutions.

RESULTS

Under these chromatographic conditions, L-glutamine was separated from nicotinic amide and calcium pantothenate; these have been prescribed frequently together (Fig. 1). L-Glutamine is not very stable in water, and decomposes forming pyrrolidonecarboxylic acid²⁸ or glutamic acid²⁹ by the action of moisture, and basic compounds, such as sodium bicarbonate, calcium carbonate, magnesium carbonate, magnesium oxide etc., in preparations stored for long periods of time. L-Glutamine was easily separated from these by-products (Fig. 2). The recovery of L-glutamine was 98.5% and the coefficient of variation was 0.98% for 5 analyses. These authors believe that this chromatographic method is very useful for the determination of L-glutamine in routine preparations, because the procedure is performed easily and rapidly.

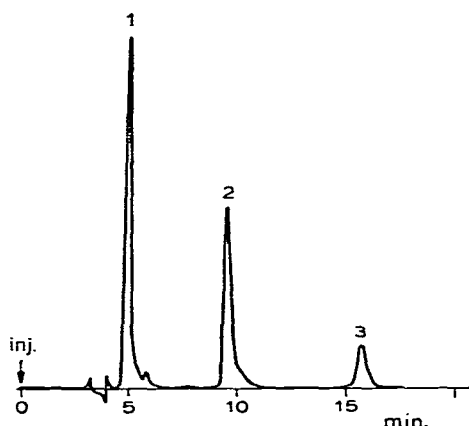


Fig. 1. Chromatogram of L-glutamine (1), nicotinic amide (2) and pantothenate (3).

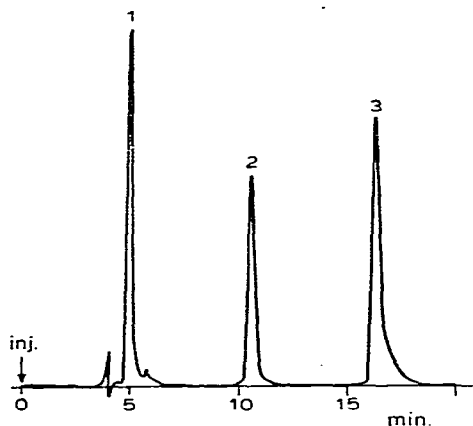


Fig. 2. Chromatogram of L-glutamine (1), pyrrolidonecarboxylic acid (3) and glutamic acid (2).

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